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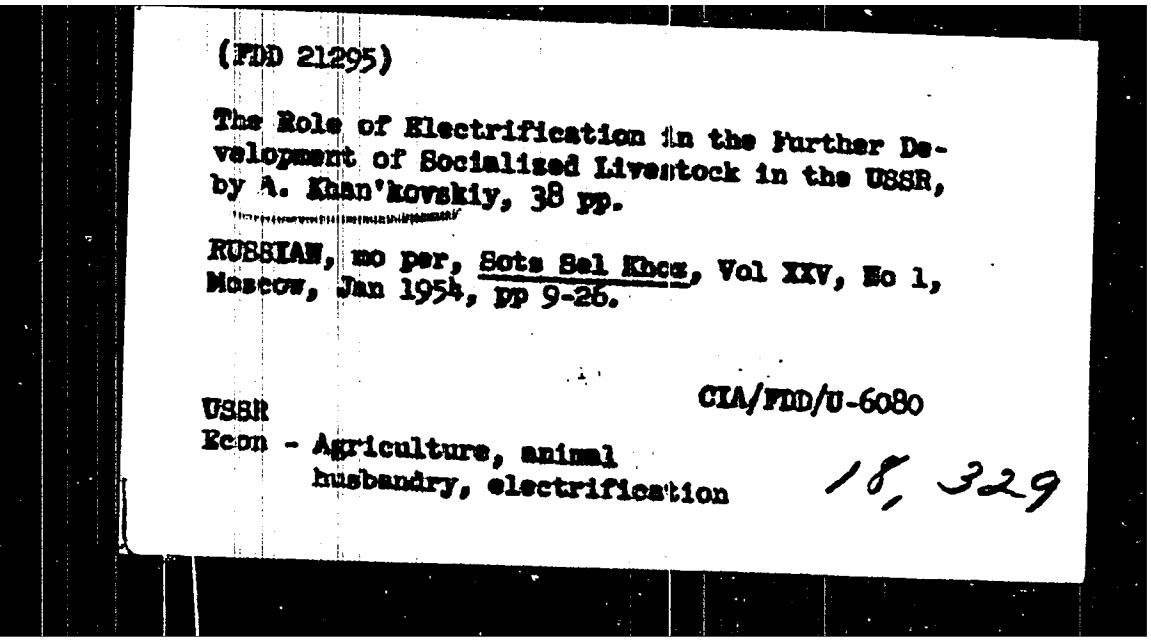

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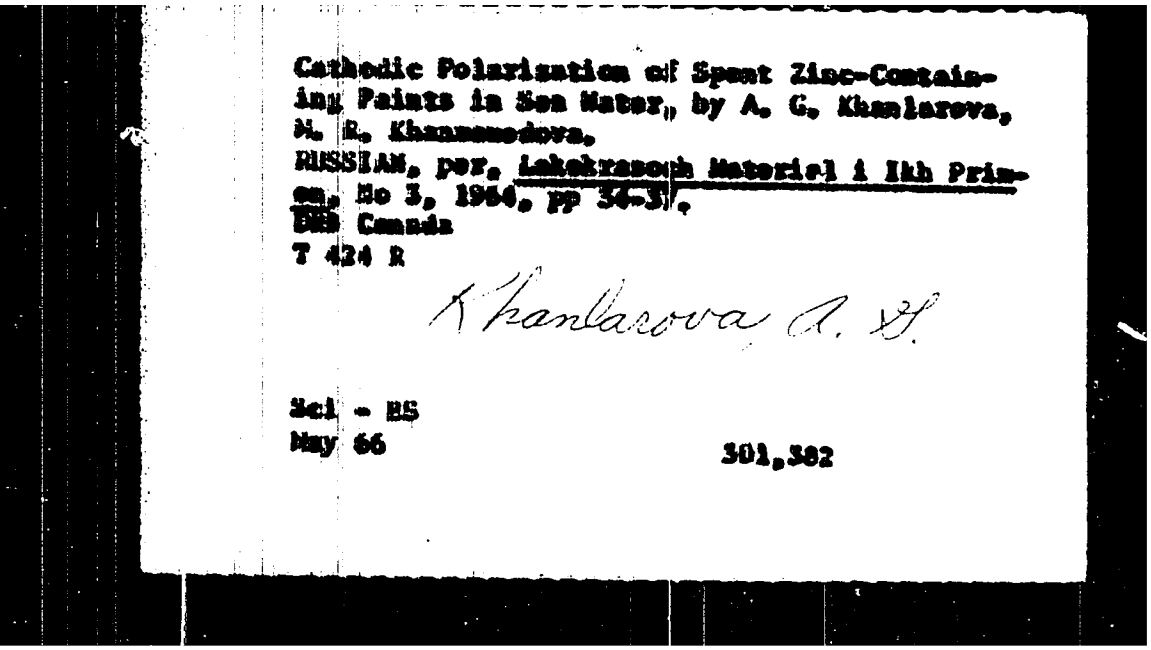

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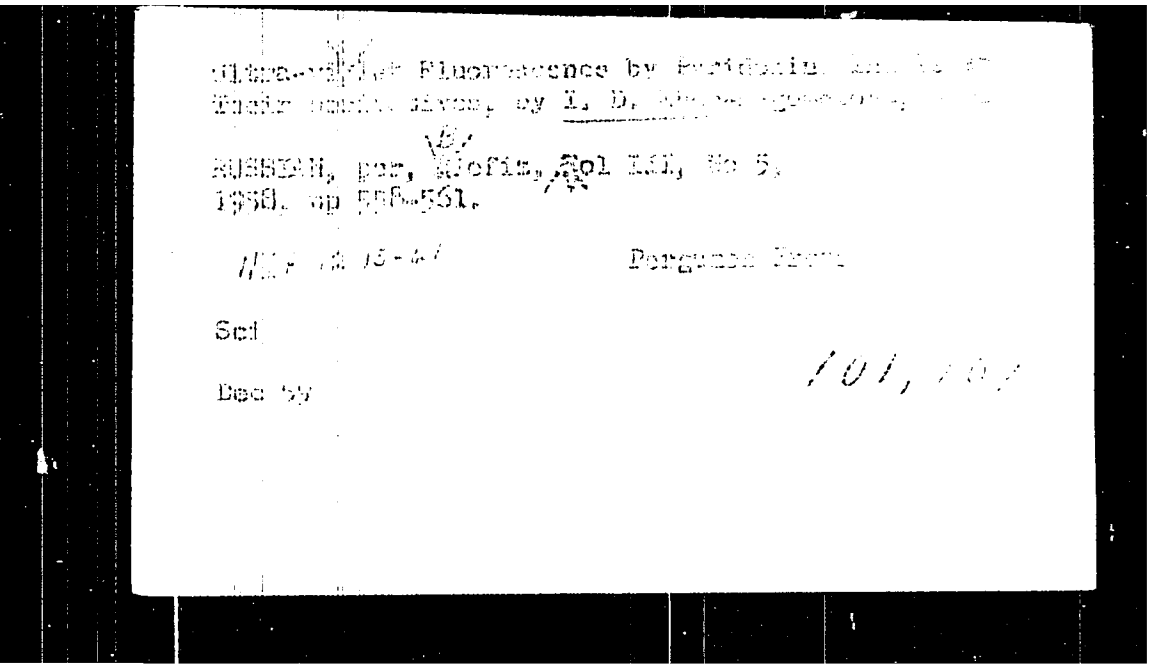

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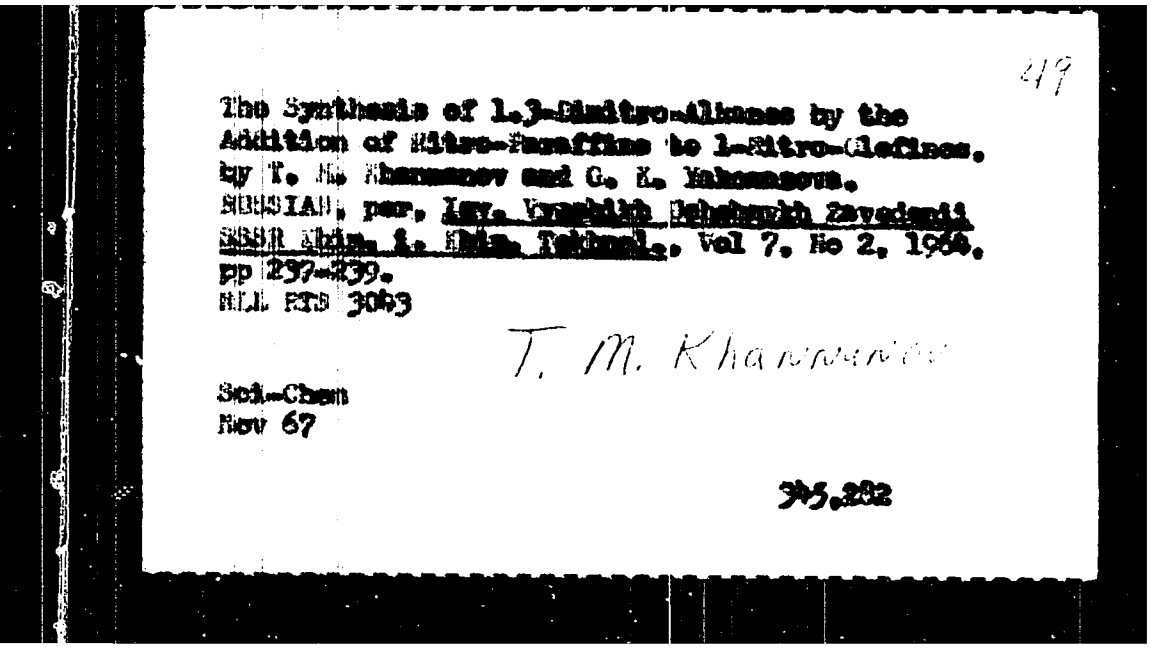

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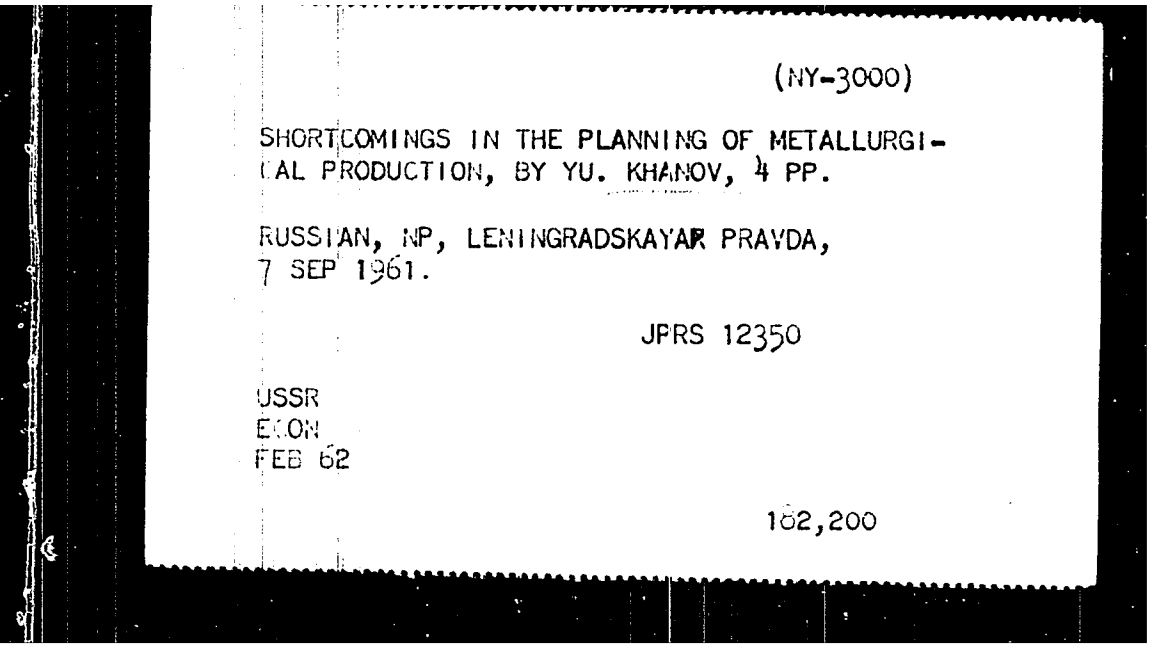

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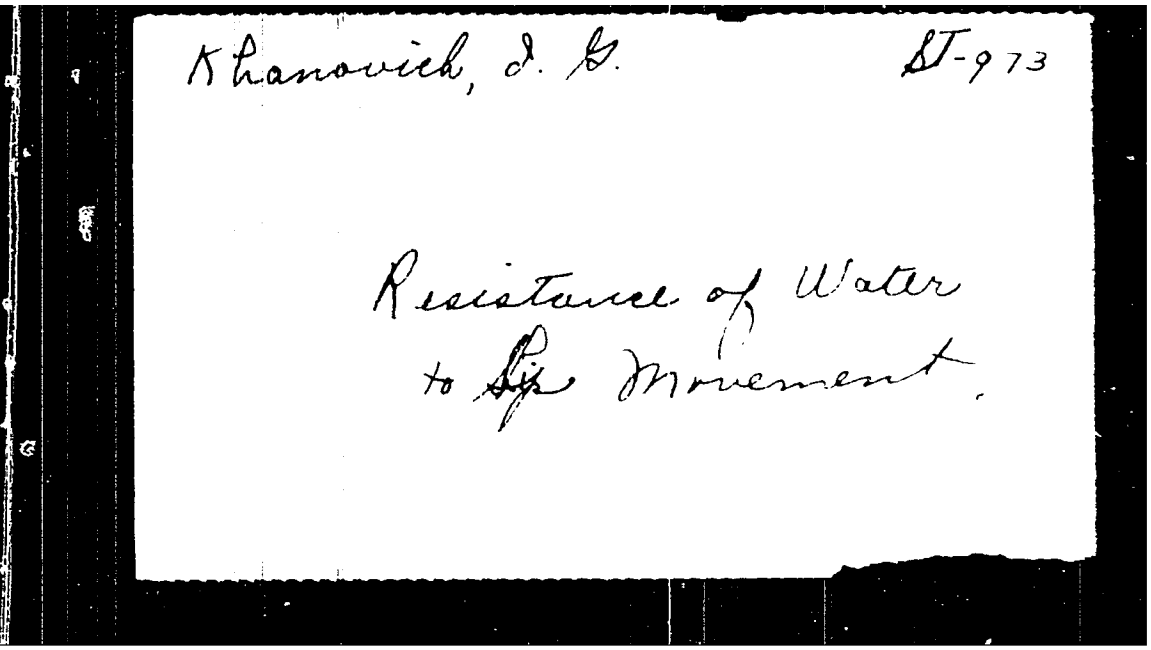

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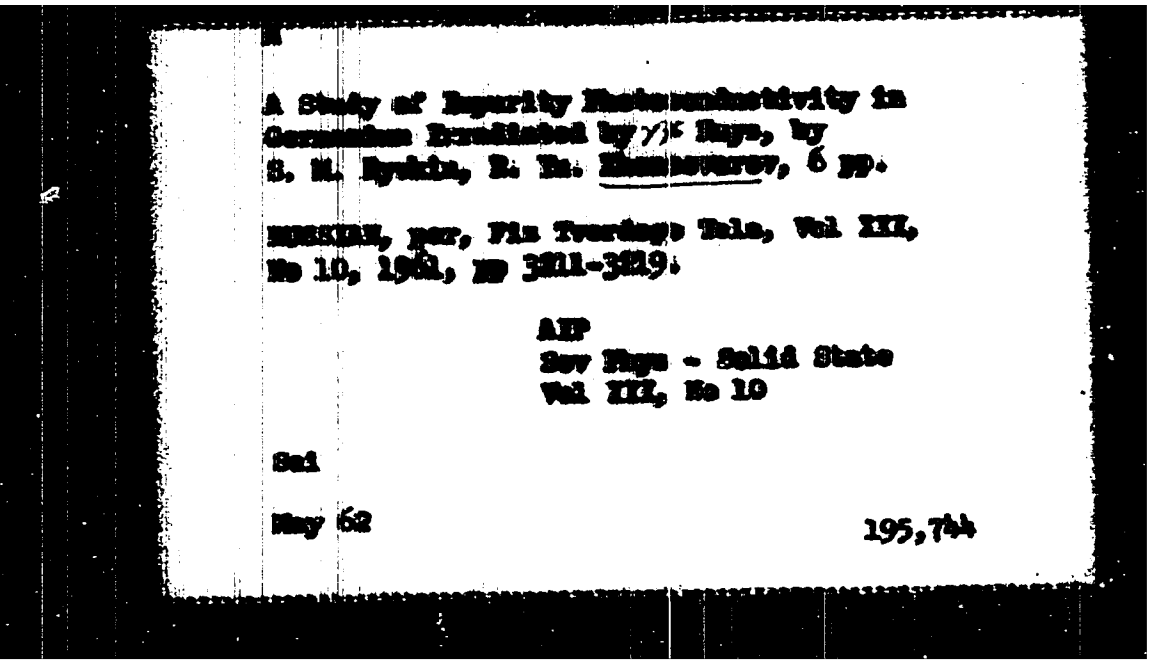

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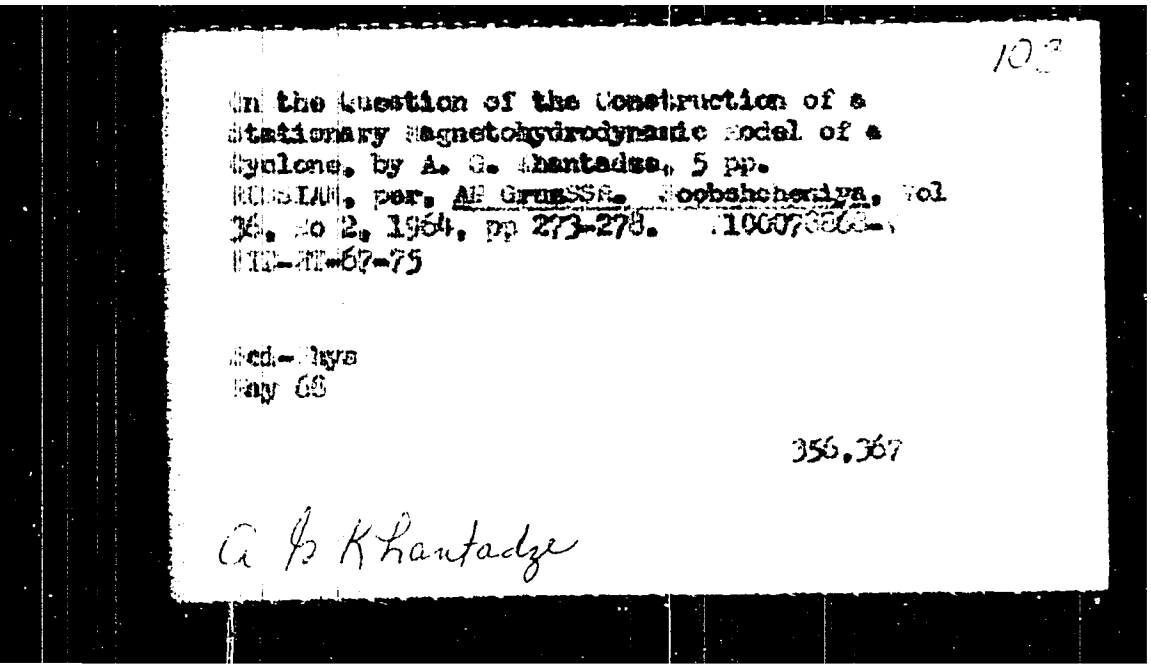

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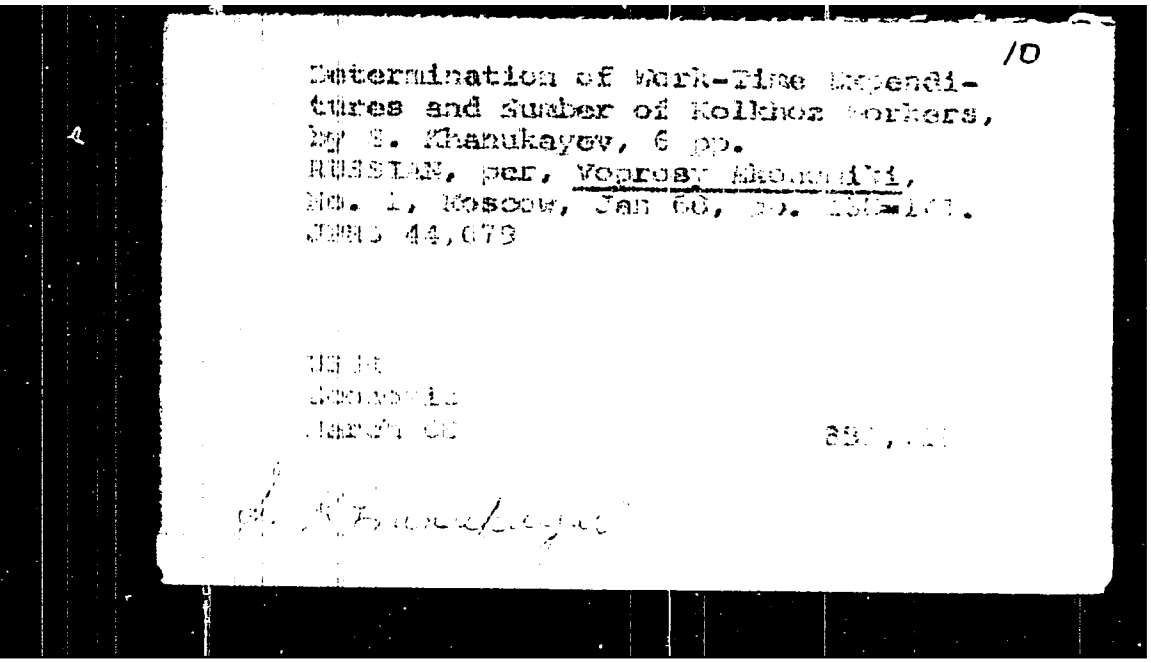

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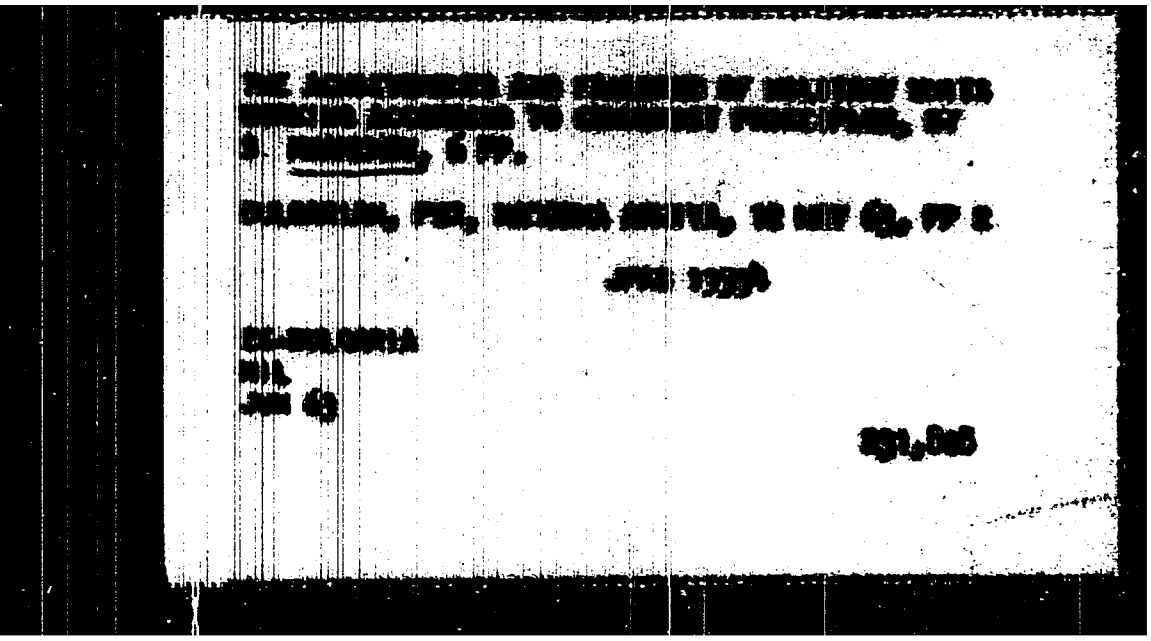

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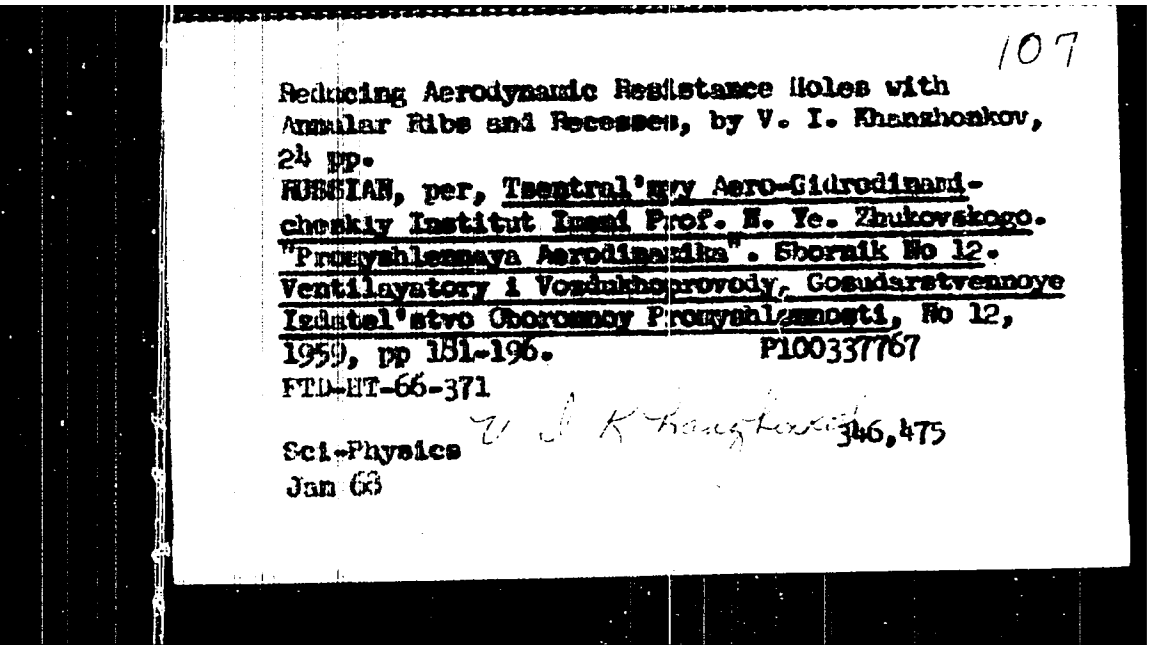

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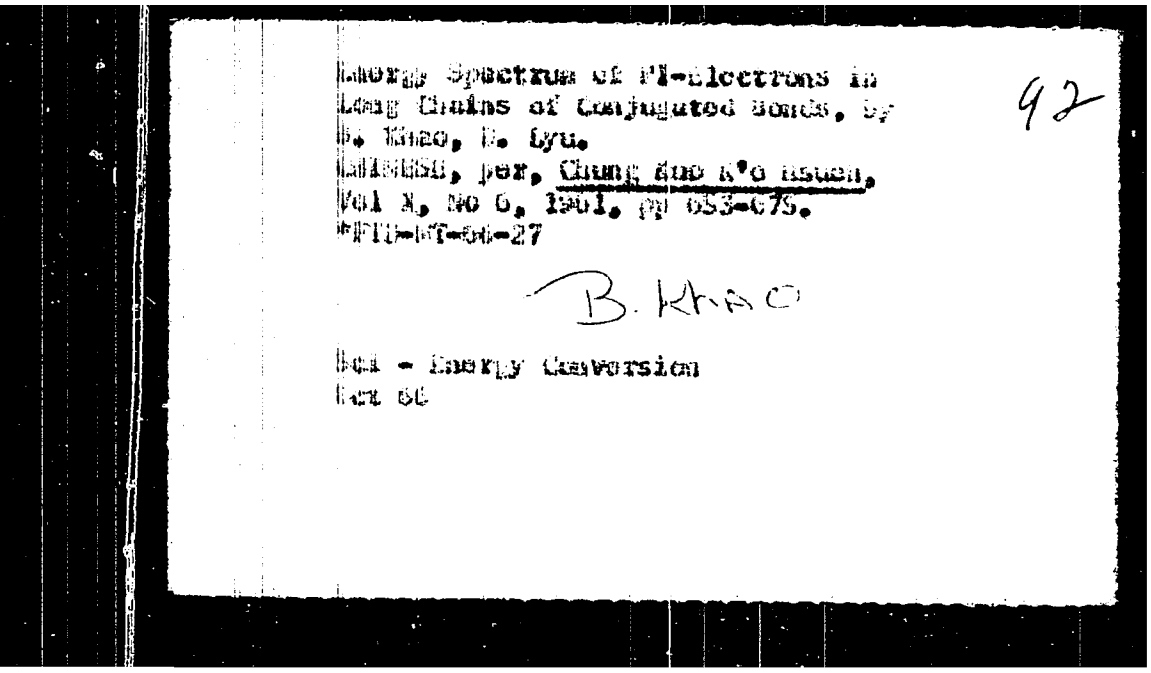

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APPROVED FOR RELEASE: Wednesday, May 29, 2002 CIA-RDP91-00772R000201150053-1

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Electromagnetic Field Inside a Plane-Parallel Plate in a Resonance Absorption Regime, by A. P. Khayaliuk.

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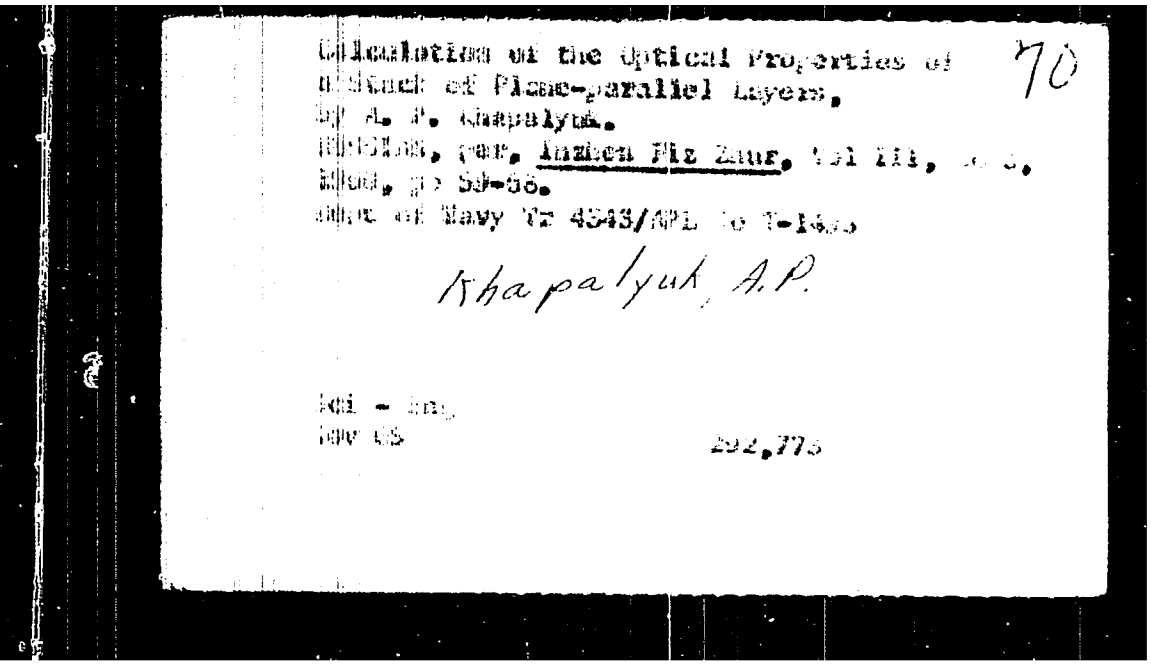

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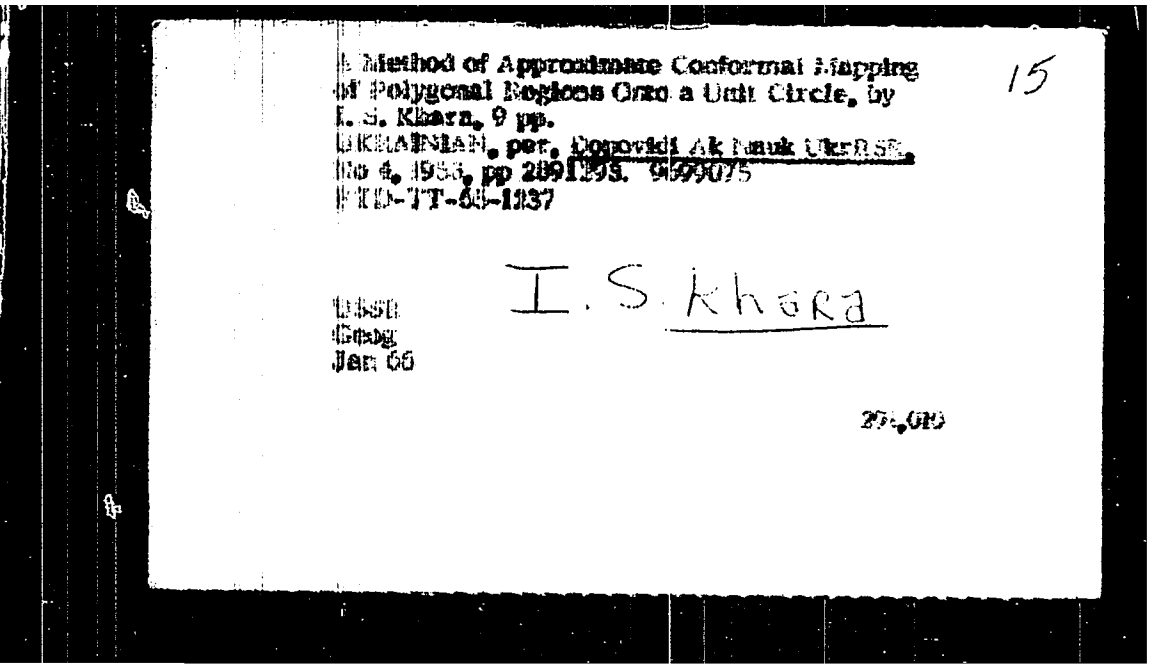

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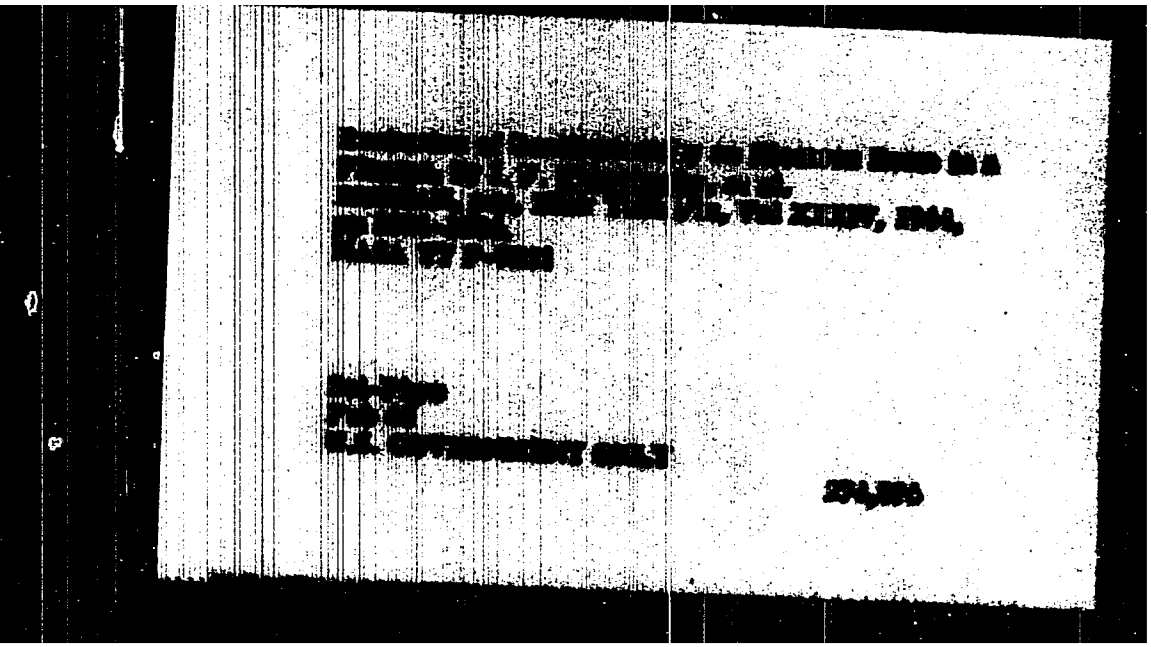

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LIQUID-VAPOR EQUILIBRIUM IN THE SYSTEMS NITROGEN-HELIUM AND HELIUM-METHANE. [1961]
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DESCRIPTORS: *Nitrogen, *Helium, *Methane, Thermodynamics, *Liquids, *Gases, Solubility.

The volatility of helium in nitrogen-helium and helium-methane systems in the temperature interval from 68 to 150.3 K and at pressures of from 4 to 215 atm is calculated from data obtained in an experimental investigation of liquidvapour equilibrium in these mixtures. It is shown that at pressures between 40 to 45 and 215 atm the solubility of helium in liquid nitrogen and methane can be calculated with sufficient accuracy from the Krichevsky--Kazarnovsky equation. The
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PHASE RELATIONSHIPS IN COMPRESSED GAS SYSTEMS. II. EQUILIBRIUM IN COEXISTING LIQUID AND VAPOR PHASES IN THE SYSTEM ETHANE-ETHYLENE. [1961] 5p. 3 refs.
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Trans. of Zhurnal Tekhnicheskoi Fiziki (USSR) 1941. v. 11, p. 1133-1139.

DESCRIPTORS: *Ethanes, *Ethylenes, Liquids, Vapors, Phase studies, Chemical equilibrium, Determination.

Equilibrium compositions of the liquid and vapor phases in the binary system ethane-ethylene were determined by the circulation method for temperatures from 169.3°K to 273.15°K under pressures from 0.5 to 40 atm and p-x and T-x diagrams constructed. It was shown that the data for the liquid and vapor phases may
(Chemistry--Physical, TT, v. 6, no. 9) (over)

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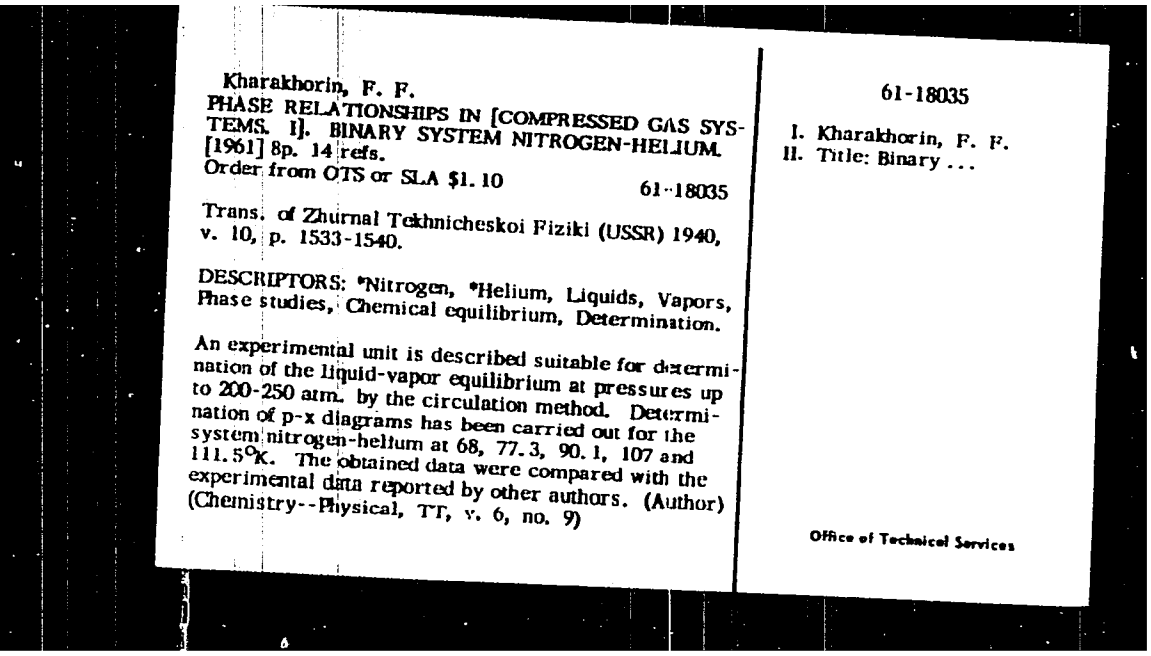

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PHASE RELATIONSHIPS IN [COMPRESSED GAS SYSTEMS. I]. BINARY SYSTEM NITROGEN-HELIUM. [1961] 8p. 14 refs.
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Trans. of Zhurnal Tekhnicheskoi Fiziki (USSR) 1940, v. 10, p. 1533-1540.

DESCRIPTORS: *Nitrogen, *Helium, Liquids, Vapors, Phase studies, Chemical equilibrium, Determination.

An experimental unit is described suitable for determination of the liquid-vapor equilibrium at pressures up to 200-250 atm. by the circulation method. Determination of p-x diagrams has been carried out for the system nitrogen-helium at 68, 77.3, 90.1, 107 and 111.5°K. The obtained data were compared with the experimental data reported by other authors. (Author) (Chemistry--Physical, TT, v. 6, no. 9)

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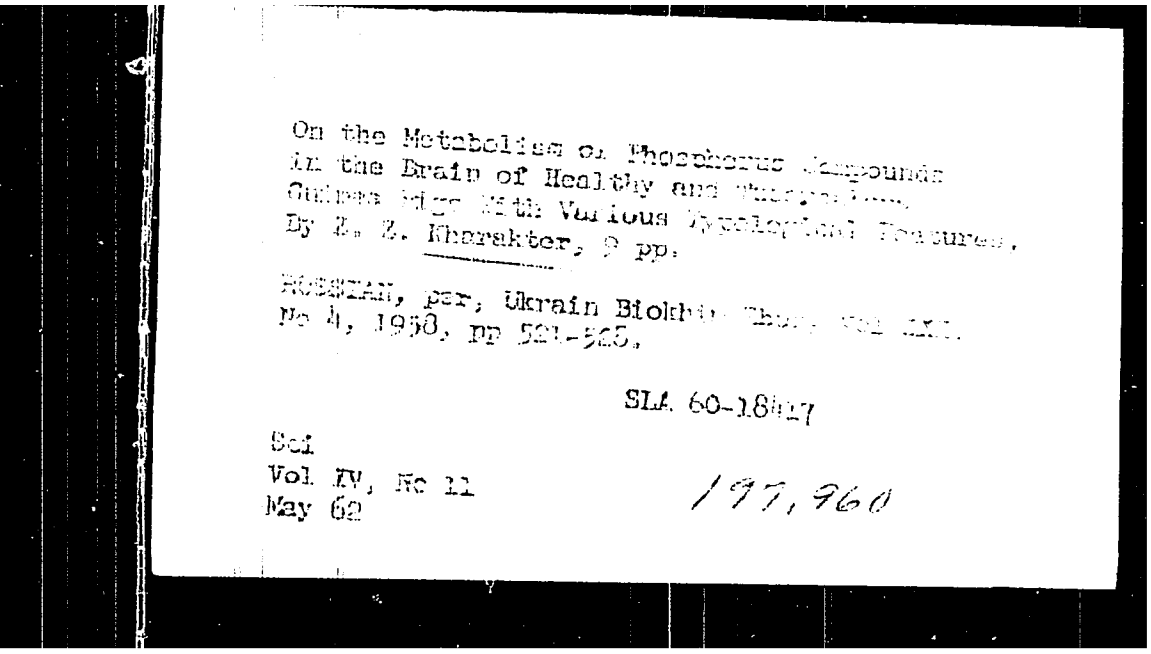

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Production of Lead in the Lead-Zinc Plant at Kurdzhali, by N. Kunchev, Ya. Yanchev, G. Kharalampiyev, 17 pp.

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Determination of the Compatibility of the Blood of Donor and Recipient by the Method of Electrophoresis of the Red Cells, by S. S. Kharamonenko, 5 pp.

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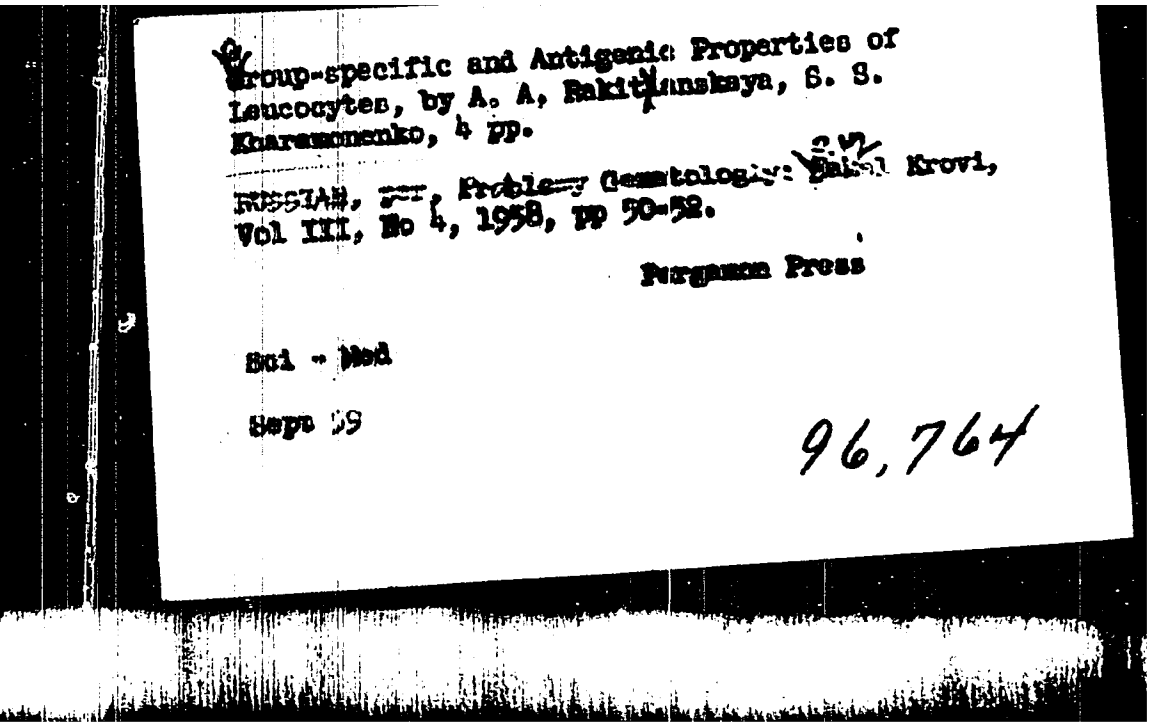

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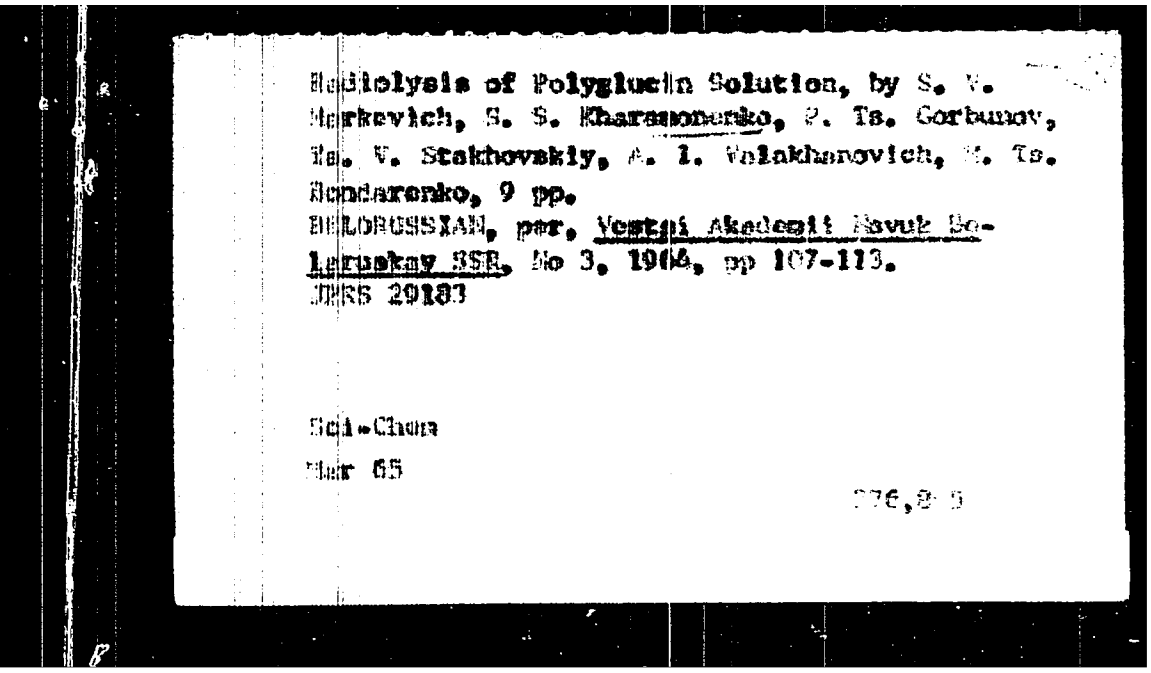

Radiolysis of Polyglucin Solution, by S. V. Markevich, S. S. Kharamonenko, P. Ts. Gorbunov, Ye. V. Stakhovskiy, A. I. Volkhonovich, M. Ts. Kondarenko, 9 pp.
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Sound Propagation in a Medium With Random Fluctuations of Refractive Index, by V. Ya. Kharunen

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On the Spin-Spin Relaxation Time in Nickel Fluosilicate in the Absence of a Static Magnetic Field by V. Ya. Kharanen, 6 pp.

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Pecularities of Respiration and Redox Processes in Rice Plants Grown Under Different Nutritional Conditions, per, N. Kharanyan.
RUSSIAN, per, Doklady Akademii Nauk SSSR
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Peculariities of Respiration and Redox Processes in Rice Plants Grown under Different Nutritional Conditions, by N. Kharanyan.
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Problems in Sinter Production, by Kharash et al.
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Q Fever Outbreak in the Kamensk Oblast, by V. P. Romanova, I. N. Petrovskiy, A. G. Somova, T. A. Nikol'skaya, P. V. Shmatko, A. A. Kosenko, V. I. Balbanova, V. G. Liparskaya, M. A. Kharathiyan, and E. M. Kompanets, 4 pp.

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The Maximum Principle in the Theory of Optimum Processes With Delay, by G. L. Kharatishvili, 5 pp.

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The Transmission of Detonation Between Initiating Substances. II. The Dependence of the Probability of Transmission of Detonation on the Distance Between the Charges and on the Area of the Passive Charge, by A. F. Belyayev, E. Mul'tovskaya, Yu. B. Kharaton, 11 pp.

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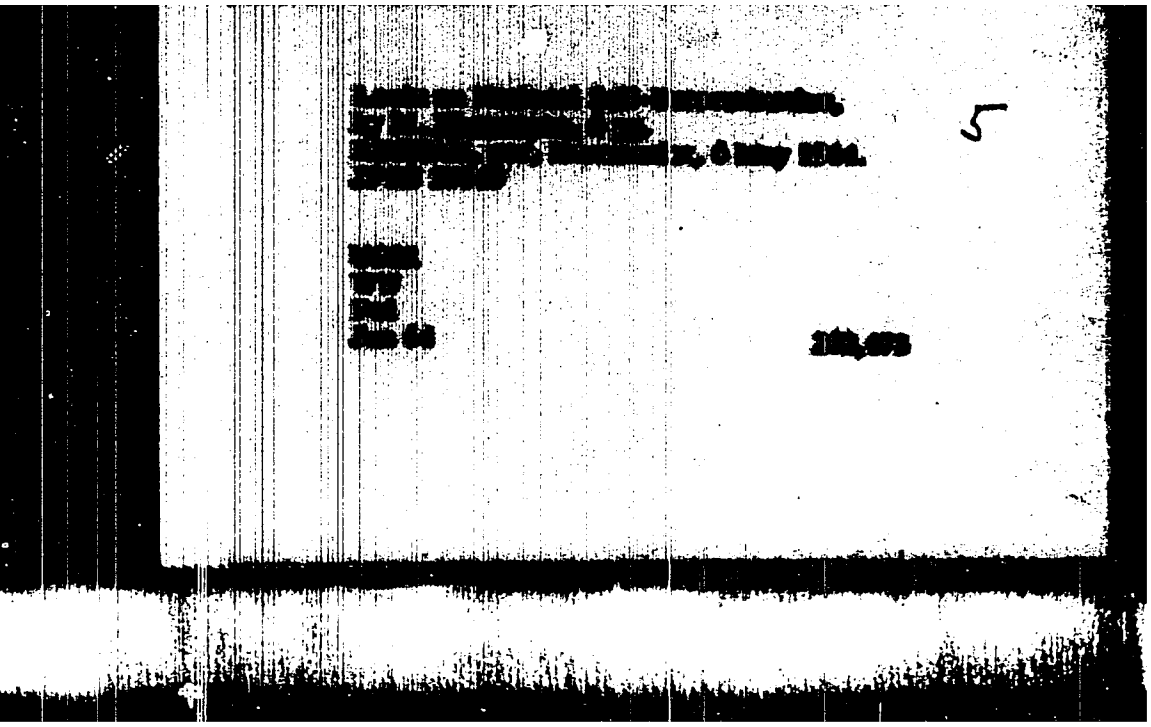

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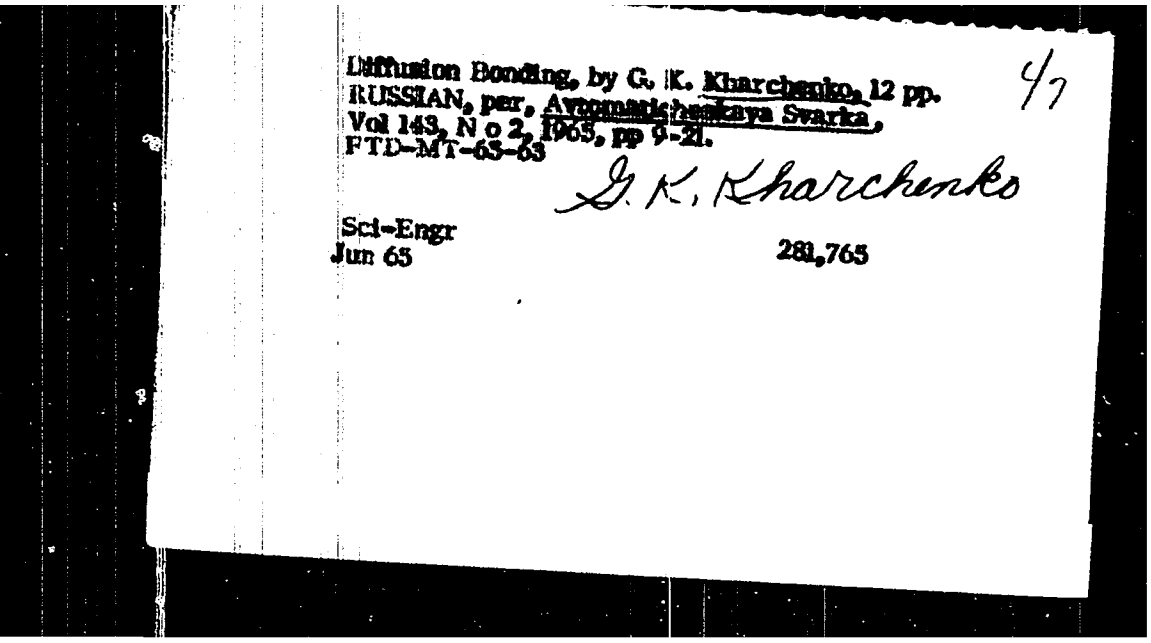

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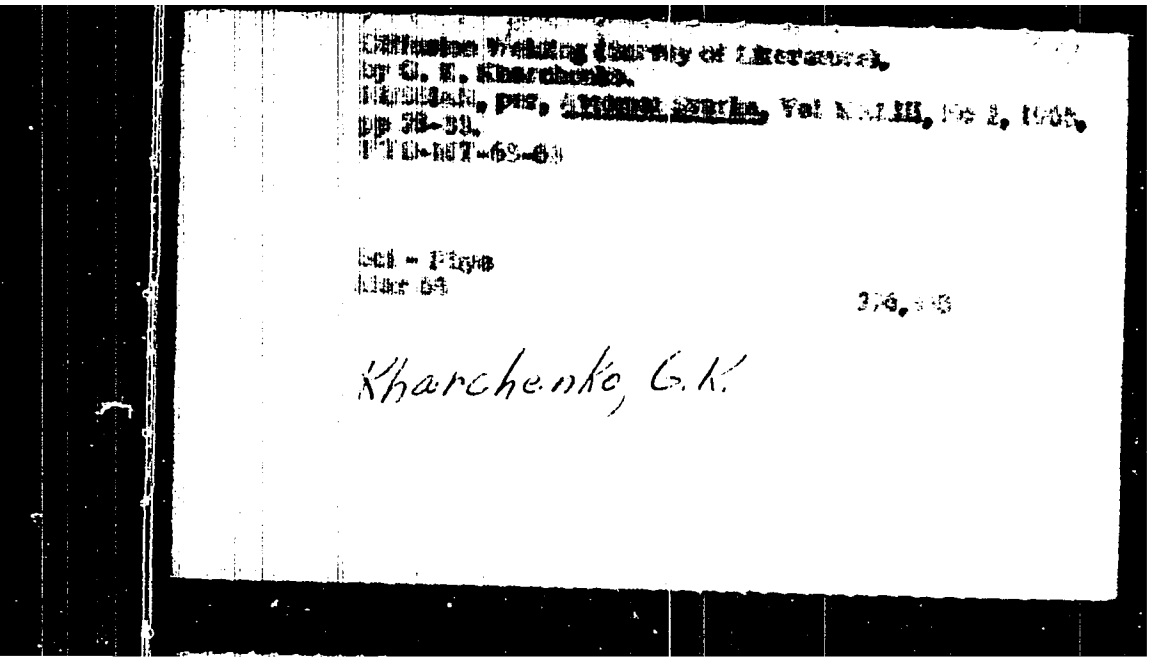

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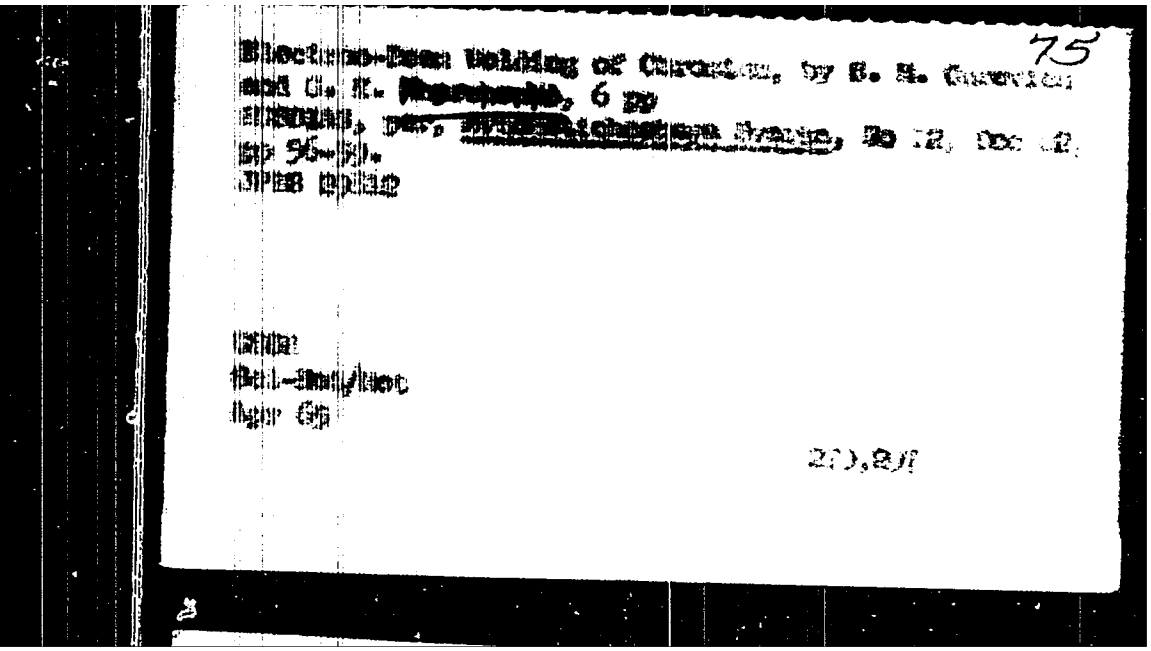

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